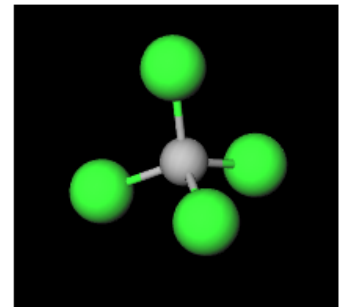
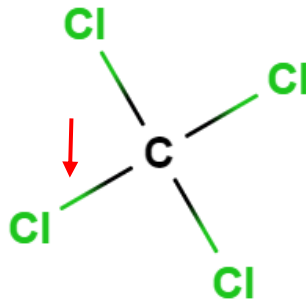


MONISTE: Kovalenttisen sidoksen poolisuus, Molekyylin poolisuus, molekyylien sitoutuminen

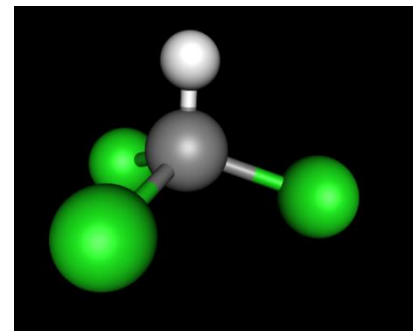
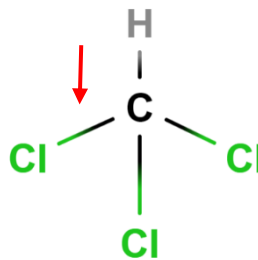
Alla on esitetty kahdeksalle eri molekyylille rakennekaava ja kolmiulotteinen rakennemalli. Vastaa jokaisen molekyylin kohdalla seuraaviin kysymyksiin:

- 1) Onko molekyylin rakennekaavaan nuolella ↓ merkitty kovalenttinen sidos poolinen vai pooliton?
- 2) Onko molekyyli kokonaisuutena poolinen vai pooliton?
- 3) Millaisilla sidoksilla saman aineen molekyylit sitoutuvat toisiinsa

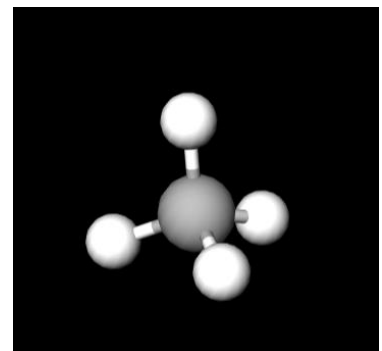
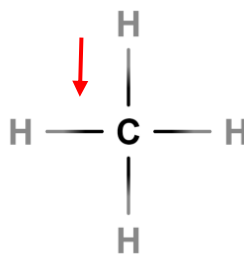
a) Tetrakloorimetaani CCl_4



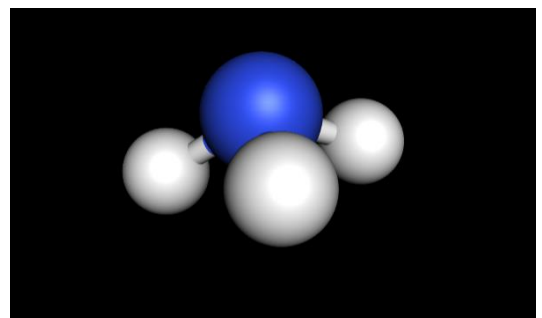
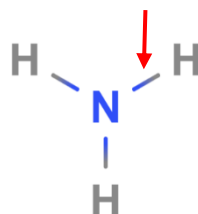
b) Trikloorimetaani CHCl_3



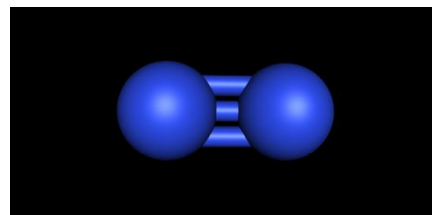
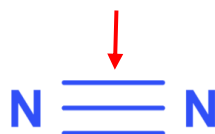
c) Metaani CH_4



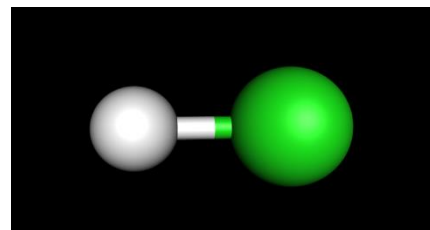
d) Ammoniakki NH_3



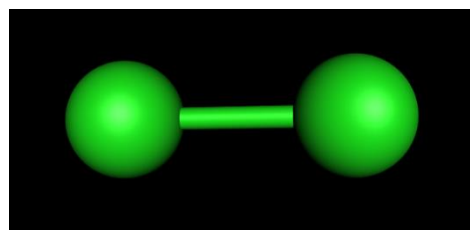
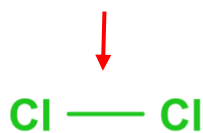
e) Typpi N_2



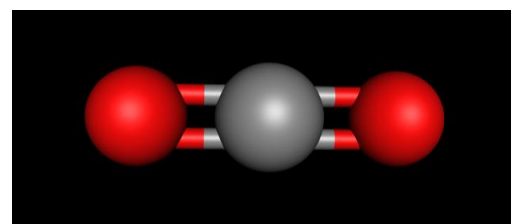
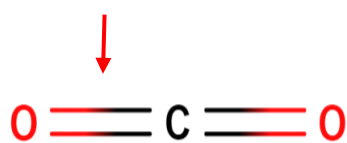
f) Vetykloridi HCl



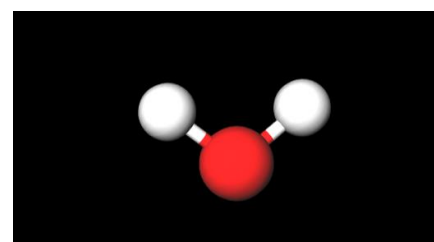
g) Kloori Cl_2



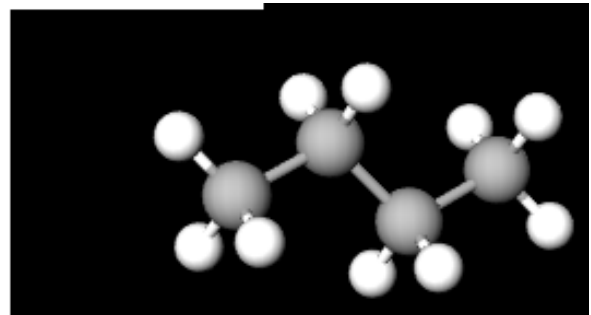
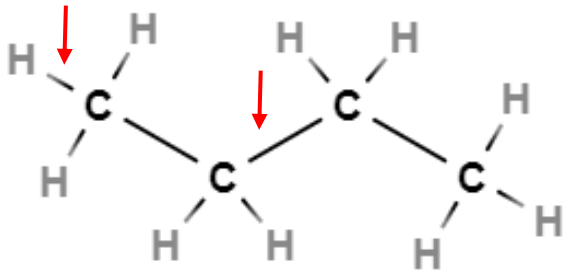
h) Hiilidioksidi CO_2



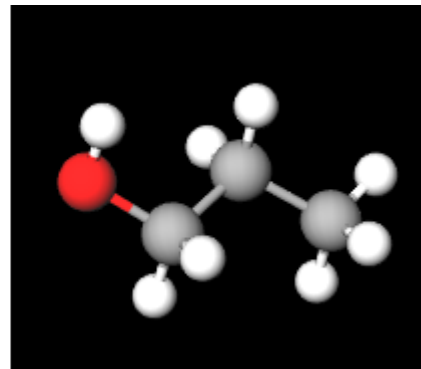
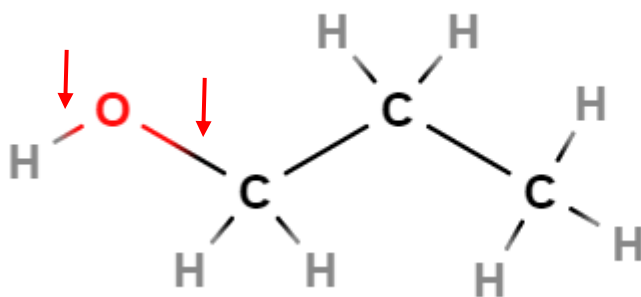
i) Vesi H_2O



j) Butaani $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3$



k) Propanoli $\text{CH}_3\text{CH}_2\text{CH}_2\text{OH}$



l) Propanihappo $\text{CH}_3\text{CH}_2\text{COOH}$

