

$$\Rightarrow \boxed{T_{1/2} = \frac{\ln 2}{\lambda}} \quad \text{PUOLINTUMISAIKA}$$

$$N = N_0 e^{-\lambda t} = N_0 e^{-\frac{\ln 2}{T_{1/2}} t} = N_0 \left(e^{\ln 2^{-1}} \right)^{\frac{t}{T_{1/2}}} = N_0 \left(\frac{1}{2} \right)^{\frac{t}{T_{1/2}}}$$

$$\Rightarrow \boxed{N = N_0 \left(\frac{1}{2} \right)^{\frac{t}{T_{1/2}}}}$$

Aktivisuus A

-ilmoitellaan montako ydin hajoaa sekunnissa

$$A = \left| \frac{\Delta N}{\Delta t} \right| = - \frac{\Delta N}{\Delta t} = \lambda N = \underbrace{\lambda N_0}_{A_0} e^{-\lambda t} = A_0 e^{-\lambda t} = A_0 \left(\frac{1}{2} \right)^{\frac{t}{T_{1/2}}}$$

$$\boxed{\begin{aligned} A &= \lambda N \\ A &= A_0 \left(\frac{1}{2} \right)^{\frac{t}{T_{1/2}}} \end{aligned}}$$

AKTIIVISUUS

$$[A] = \frac{1}{s} = \text{Bq} \quad (\text{Becquerel})$$

K25. a) Puolintumisaika $T_{1/2}$ on hyvin lyhyt (25,5h ... 75 432,6a) muille isotooppeille verrattuna ^{232}Th -isotooppiin $1,405 \cdot 10^{10}$ a. Siten muut isotoopit ovat ennustettavissa jo hetimit.



$$c) \quad \lambda = \frac{\ln 2}{T_{1/2}} = \frac{\ln 2}{1,405 \cdot 10^{10} \cdot 365,25 \cdot 24 \cdot 60 \cdot 60 \text{ s}} = 1,563 \cdot 10^{-18} \frac{1}{s}$$

$$d) \quad A = \lambda N = \lambda \frac{m}{m(^{232}\text{Th})} = 1,563 \cdot 10^{-18} \frac{1}{s} \cdot \frac{2,0 \cdot 10^{-3} \text{ kg}}{232,038 \cdot 1,66054 \cdot 10^{-27} \text{ kg}} \\ \approx 8114,59 \frac{1}{s} \Rightarrow \text{N. 8100 ydintä-}$$

$$e) \quad N = N_0 \left(\frac{1}{2} \right)^{\frac{t}{T_{1/2}}} = 0,8 N_0 \quad | : N_0 \\ \Rightarrow \left(\frac{1}{2} \right)^{\frac{t}{T_{1/2}}} = 0,8 \quad | \ln \quad \leftarrow \text{hajonnut 20\%} \rightarrow \text{jäljellä 80\%}$$

$$\Rightarrow \frac{t}{T_{1/2}} \ln \frac{1}{2} = \ln 0,8 \quad | \cdot \frac{T_{1/2}}{\ln \frac{1}{2}}$$

$$\Rightarrow t = \frac{\ln 0,8 \cdot T_{1/2}}{\ln \frac{1}{2}} = \dots = 4,5 \cdot 10^9 \text{ a} \quad \approx \text{aurinkokunnan ikä}$$

$$f) \quad A = 0,8 A_0 = 0,8 \cdot 8114,59 \text{ Bq} = \underline{6,5 \text{ kBq}}$$